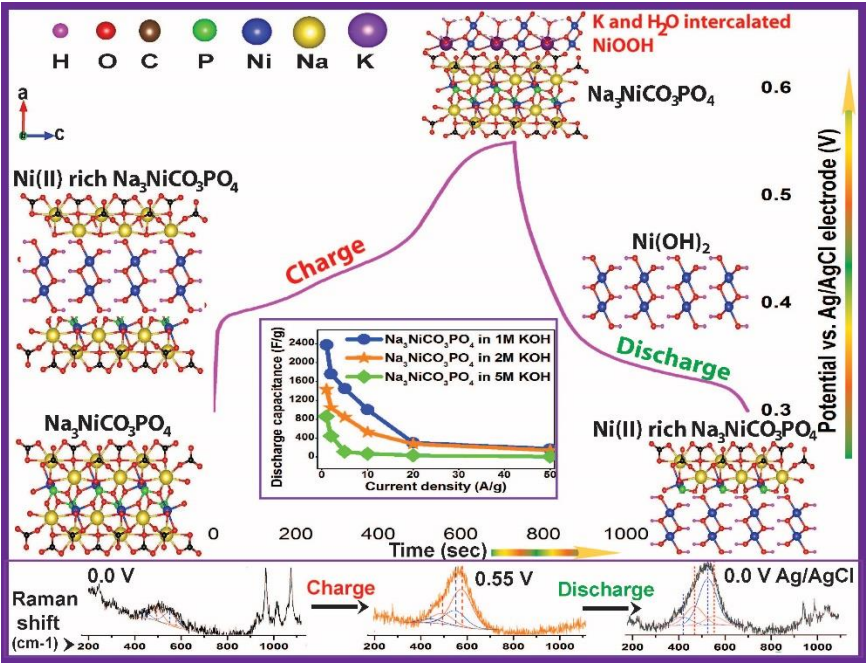


**Website Information**  
**Department of Physics**  
**Dayananda Sagar University**

**Faculty details:**

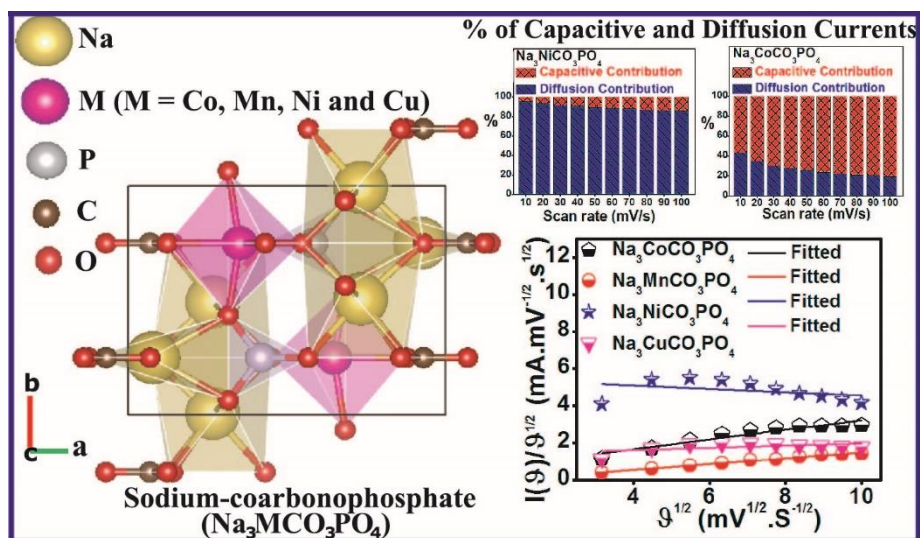
Item	Details
Faculty Name:	Dr. Yogesh Kalegowda
Room No:	A-213
Designation:	Assistant Professor
E-Mail	yogesh-phy@dsu.edu.in
Research Area:	<p>Syncrotron Soft X-ray Spectroscopy and Spectro-microscopy</p> <p>Energy Storage and Conversion Materials</p> <p>Energy Storage Devices</p> <p>Artificial Intelligence and Machine Learning</p>
Publications (Past 5 years)	<ol style="list-style-type: none"> <li>1. B.S. Nishchith, Yogesh Kalegowda*, S. Ashoka, L. Shreenivasa, Ganesan Sriram, Mahaveer D. Kurkuri, Kanalli Vinayak Ajeya, Ho-Young Jung, Electrochemical kinetic study and performance evaluation of surface-modified mesoporous sodium carbonophosphates nanostructures for pseudocapacitor applications, Journal of Alloys and Compounds, 939, 2023, 168711.</li> <li>2. P. Sagar, K. Yogesh, Asad Syed, Najat Marraiki, Abdallah M. Elgorban, Nouf S. S. Zaghoul, S. Ashoka, Studies on the effect of crystalline Fe<sub>2</sub>O<sub>3</sub> on OER performance of amorphous NiOOH electrodeposited on stainless steel substrate, Chemical Papers 76 (11), 2022, 7195-7203.</li> <li>3. Srinivasa, N., Shreenivasa, L., Ashoka, S., &amp; Yogesh, K, Studies on Co<sub>3</sub>O<sub>4</sub>-NiO nanocomposites for potential electrocatalyst for alkaline water electrolysis. Applied Physics A: Materials Science &amp; Processing, 128(2), 2022, 158.</li> <li>4. Shreenivasa, L., Ashoka, S., Yogesh, K., Syed, A., Marraiki, N., &amp; Adarakatti, P. S., Glycine-nitrate derived cobalt-doped BiPO<sub>4</sub>: An efficient OER catalyst for alkaline electrochemical cells. Solid State Sciences, 124, 2022, 106803.</li> <li>5. Nishchith, B.S., Ashoka, S., Bhat, M.P., Kurkuri, M.D., Acharya, S., Kumar, R. and Kalegowda, Y*, Reversible surface reconstruction of Na<sub>3</sub>NiCO<sub>3</sub>PO<sub>4</sub>: A battery type electrode for pseudocapacitor applications. Journal of Power Sources, 520, 2022, 230903.</li> </ol>

	<ol style="list-style-type: none"> <li>6. Shreenivasa, L., Yogeeshwari, R.T., Viswanatha, R., Yogesh, K. and Ashoka, S., Sucrose-assisted rapid synthesis of multifunctional <math>\text{CrVO}_4</math> nanoparticles: a new high-performance cathode material for lithium ion batteries. <i>Ionics</i>, 27(1), 2021, 39-48.</li> <li>7. Shreenivasa, L., Viswanatha, R., Ganesan, S., Kalegowda, Y., Kurkuri, M.D. and Ashoka, S., Scalable chemical approach to prepare crystalline <math>\text{Mn}_2\text{V}_2\text{O}_7</math> nanoparticles: introducing a new long-term cycling cathode material for lithium-ion battery. <i>Journal of Materials Science: Materials in Electronics</i>, 31(22), 2020, 19638-19646.</li> <li>8. Shreenivasa, L., Yogesh, K., Prashanth, S.A., Viswanatha, R. and Ashoka, S., Enhancement of cycling stability and capacity of lithium secondary battery by engineering highly porous <math>\text{AlV}_3\text{O}_9</math>. <i>Journal of Materials Science</i>, 55(4), 2020, 1648-1658.</li> <li>9. Shreenivasa, L., Yogeeshwari, R.T., Viswanatha, R., Sriram, G., Kalegowda, Y., Kurkuri, M.D. and Ashoka, S., An introduction of new nanostructured <math>\text{Zn}_{0.29}\text{V}_2\text{O}_5</math> cathode material for lithium ion battery: a detailed studies on synthesis, characterization and lithium uptake. <i>Materials Research Express</i>, 6(11), 2019, 115035.</li> <li>10. Shreenivasa, L., Prashanth, S.A., Eranjaneya, H., Viswanatha, R., Yogesh, K., Nagaraju, G. and Ashoka, S., Engineering of highly conductive and mesoporous <math>\text{ZrV}_2\text{O}_7</math>: a cathode material for lithium secondary batteries. <i>Journal of Solid State Electrochemistry</i>, 23(4), 2019, 1201-1209.</li> <li>11. Lakkepally, S., Kalegowda, Y., Ramarao, V., Hanumantharayappa, E. and Siddaramanna, A., Room temperature synthesis of amorphous <math>\text{Bi}_4\text{V}_2\text{O}_{11}</math> as cathode material for Li secondary batteries. <i>Materials Research Express</i>, 5(11), 2018, 115501.</li> <li>12. Lakkepally, S., Kalegowda, Y., Ganganagappa, N. and Siddaramanna, A., A new and effective approach for <math>\text{Fe}_2\text{V}_4\text{O}_{13}</math> nanoparticles synthesis: Evaluation of electrochemical performance as cathode for lithium secondary batteries. <i>Journal of Alloys and Compounds</i>, 737, 2018, 665-671.</li> <li>13. Varhade, S., Bhat, Z.M., Thimmappa, R., Devendrachari, M.C., Kottaichamy, A.R., Gautam, M., Shafi, S.P., Kalegowda, Y. and Thotiyl, M.O., A hybrid hydrazine redox flow battery with a reversible electron acceptor. <i>Physical Chemistry Chemical Physics</i>, 20(33), 2018, 21724-21731.</li> </ol>
Sponsored Projects (Past and Ongoing)	Graphene oxide (GO) and reduced graphene oxide (rGO) based nanomaterials for photothermal supercapacitor applications (DST-SERB project-Ongoing)

<p>Profile Links: Scopus and Orcid</p>	<p><a href="https://orcid.org/0000-0001-6956-9164">https://orcid.org/0000-0001-6956-9164</a> AAV-7055-2021 (Web of Science) <a href="https://scholar.google.com/citations?hl=en&amp;authuser=1&amp;user=ri4QDysAAAAJ">https://scholar.google.com/citations?hl=en&amp;authuser=1&amp;user=ri4QDysAAAAJ</a></p>
<p>Research Activities (Write about your best research results max of 2-3 pages including diagrams)</p>	<p><b>Reversible Surface Reconstruction of <math>\text{Na}_3\text{NiCO}_3\text{PO}_4</math>: A Battery Type Electrode for Pseudocapacitor Application</b> (<a href="https://doi.org/10.1016/j.jpowsour.2021.230903">https://doi.org/10.1016/j.jpowsour.2021.230903</a>)</p> <p>This work establishes fundamental understanding on the pseudocapacitance mechanism associated with surface self-reconstruction of mixed polyanionic compounds. The reconstruction-derived self-limiting dense <math>\text{Ni}(\text{OH})_2</math> layers at the surface and its oxidation to <math>\text{NiOOH}</math> at anodic potentials can be attributed to the observed high performance pseudocapacitive behavior. The surface reconstructed <math>\text{Na}_3\text{NiCO}_3\text{PO}_4</math> electrode exhibits high specific capacitance (<math>2378.2</math> and <math>169.5 \text{ F g}^{-1}</math> at <math>1</math> and <math>50 \text{ A g}^{-1}</math>, respectively) and long cycle life (<math>15000</math> at <math>50 \text{ A g}^{-1}</math>) in <math>1 \text{ M KOH}</math> electrolyte. The assembled symmetric pseudocapacitor delivers a highest energy density of <math>146.0 \text{ Wh kg}^{-1}</math> and power density of <math>1600 \text{ W kg}^{-1}</math> at <math>1 \text{ A g}^{-1}</math>.</p>  <p><b>Electrochemical Kinetic Study and Performance Evaluation of Surface-modified Mesoporous Sodium Carbonophosphates Nanostructures for Pseudocapacitor Applications</b></p>

(<https://doi.org/10.1016/j.jallcom.2023.168711>)

- The pseudocapacitive behavior of the surface modified  $\text{Na}_3\text{MCO}_3\text{PO}_4$  (M=Ni, Co, Mn and Cu) nanostructures is tested.
- The  $\text{Na}_3\text{MCO}_3\text{PO}_4$  (M=Co and Mn) undergo deep reconstruction in 1 M KOH to form  $\text{M}(\text{OH})_2/\text{M}_3\text{O}_4$  nanocomposite.
- The capacitive and diffusion currents contribution are clearly Identified.
- The deep reconstructed  $\text{Co}(\text{OH})_2/\text{Co}_3\text{O}_4$  symmetric supercapacitor delivers a high energy and power densities.



Group Members  
(PhD Students and Projects)

BS Nishchith, Prasanna DG, Narasimha SA

Open Positions: If any

Project Assistant-1  
Qualification: M.Sc. Physics